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# **Particle Measurement Methodology: Comparison of On-road and Lab Diesel Particle Size Distributions**

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# **This work is part of the CRC E-43 Project, “Diesel Aerosol Sampling Methodology”**

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- Prime Contractor: University of Minnesota
- Subcontractors: West Virginia University, Paul Scherrer Institute, Carnegie Mellon University, Tampere University, University of California, Riverside, Desert Research Institute, University of California, Davis
- Sponsors: Coordinating Research Council and the U.S. Office of Heavy Vehicle Technologies through NREL with co-sponsorship from the Engine Manufacturers Association, the Southcoast Air Quality Management District, the California Air Resources Board, Cummins, Caterpillar, and Volvo.

## E-43 Questions

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- Do modern Diesel engines produce nanoparticles under real world dilution conditions?
- Can we make laboratory measurements that mimic real world measurements?
- Do new low carbon emitters produce more nanoparticles than older designs?
- What is the composition of the nanoparticles?
- How long do they persist in the atmosphere?

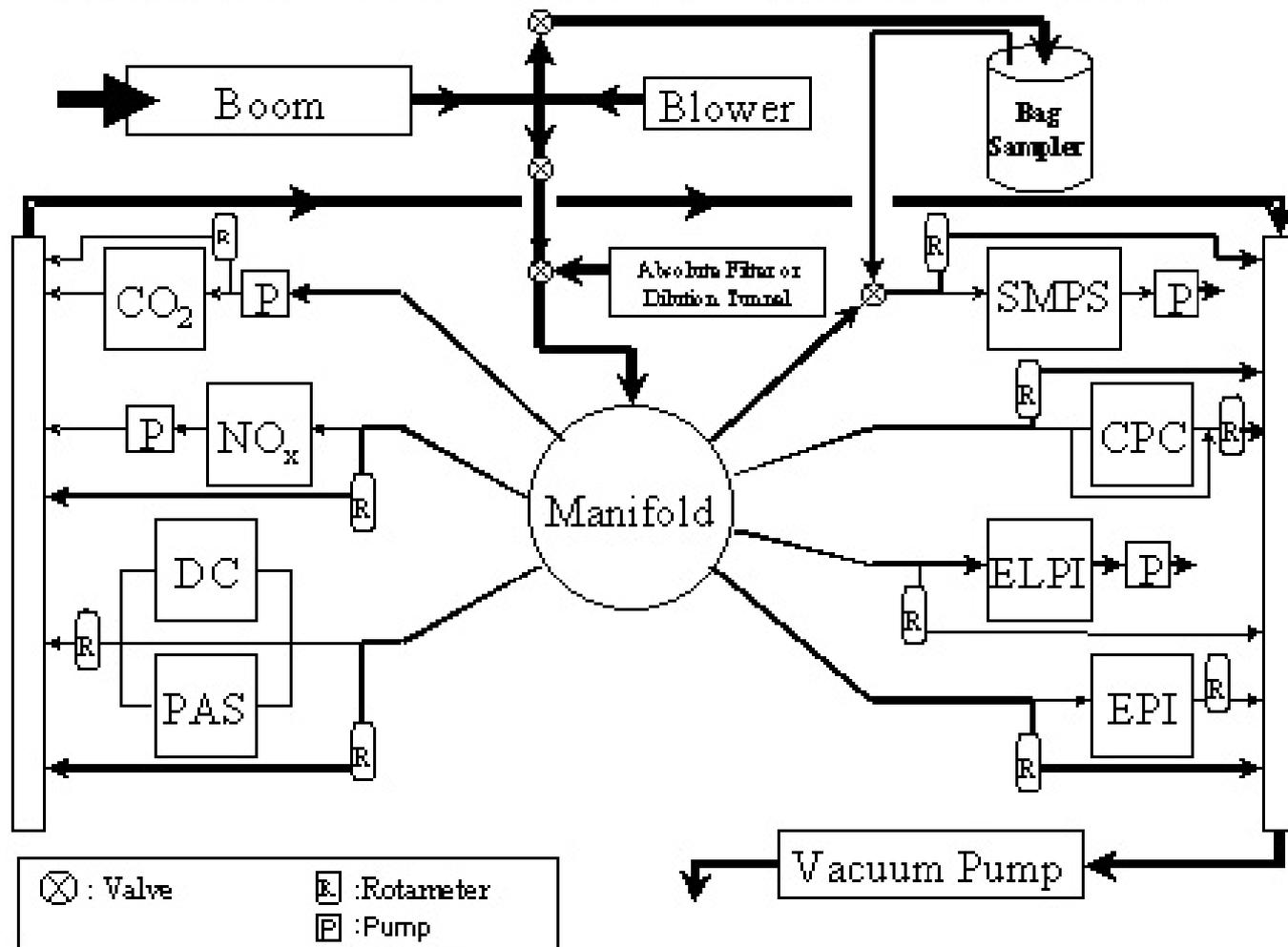
# **E-43 Experiments – current and older technology engines without aftertreatment**

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- Cummins engines
  - Chase experiments
    - ISM engine CA and EPA fuels
    - L10 engine EPA fuel
  - Wind tunnel – ISM engine CA fuel
  - Chassis dyno
    - ISM engine CA and EPA fuels
    - L10 engine EPA fuel
  - Engine dyno
    - ISM engine CA and EPA fuels
    - L10 engine EPA fuel
  - Tests of ISM engine at U of M
    - TDPBMS
    - Tandem DMA
- Caterpillar engines
  - Chase experiments
    - 3406E (C15) engine CA and EPA fuels
    - 3406C engine EPA fuel
  - Chassis dyno
    - 3406E (C15) engine CA and EPA fuels
    - 3406C engine EPA fuel
  - Engine dyno Caterpillar
    - 3406E (C15) in CVS cell
    - 2 additional 3406E in performance cell
  - Tests of C12 engine at U of M
    - Dilution system development
    - TDPBMS

# Instrument and Sampling Arrangement in Mobile Emission Laboratory

Mobile Emission Laboratory (MEL) Flow System Chart



# University of Minnesota, E-43, Mobile Aerosol Laboratory during a Roadway Chase Experiment

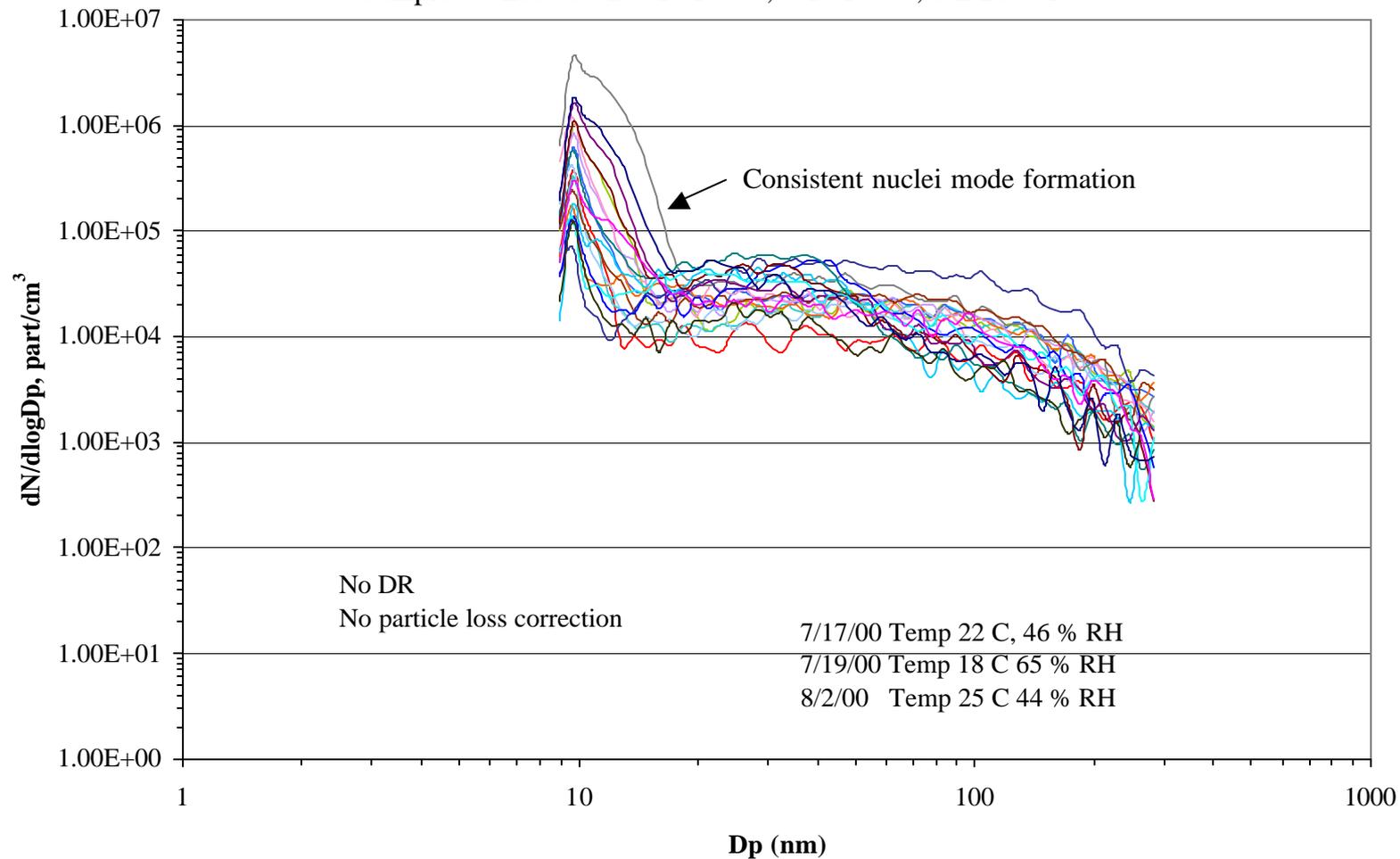
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# On-road test conditions were very unsteady even under “steady state” conditions

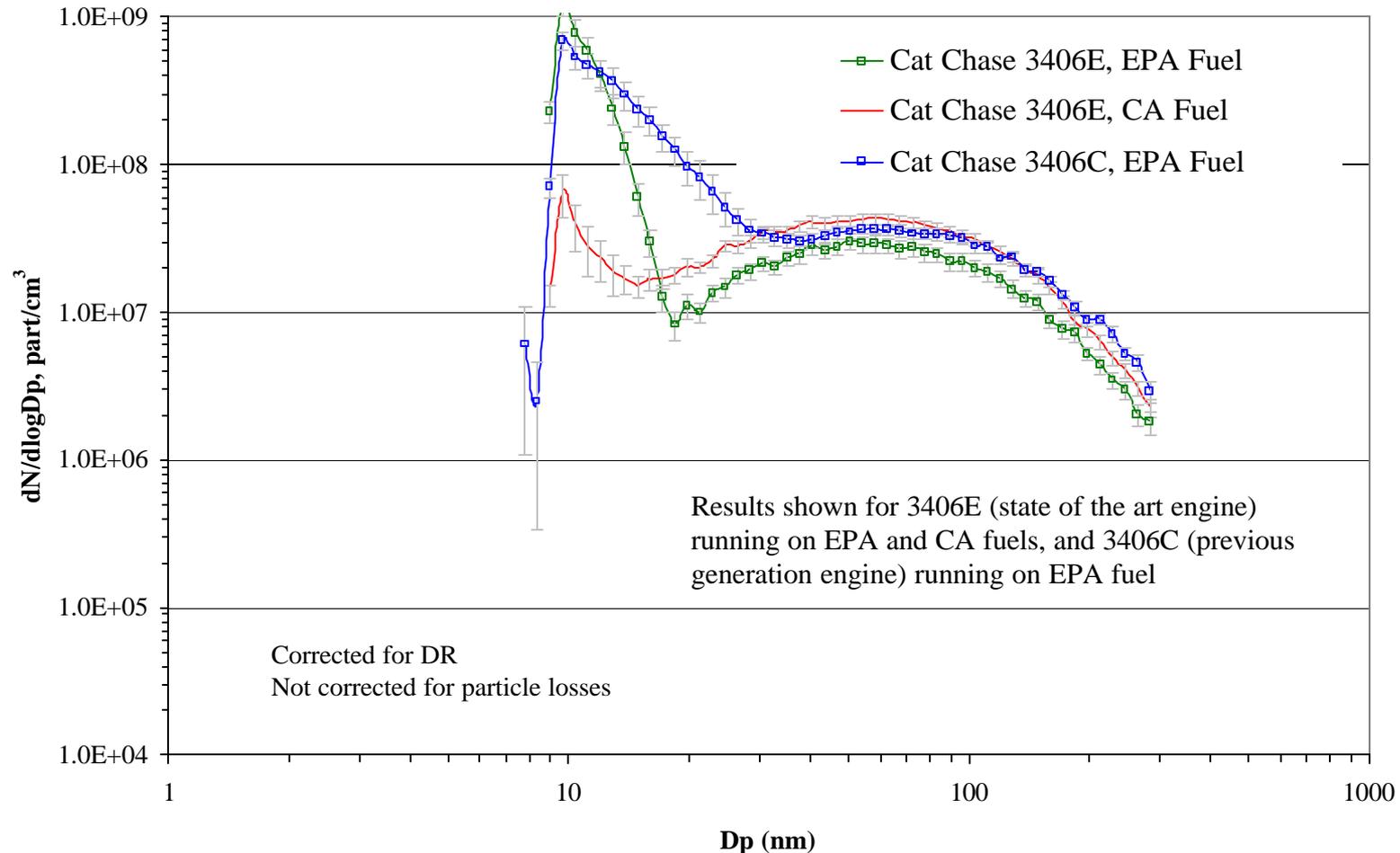
3406E Truck, Loaded, 60 MPH Cruise, EPA Fuel

Samples collected on 7/17 N = 9, 7/19 N = 5, 8/2/2000 N = 6



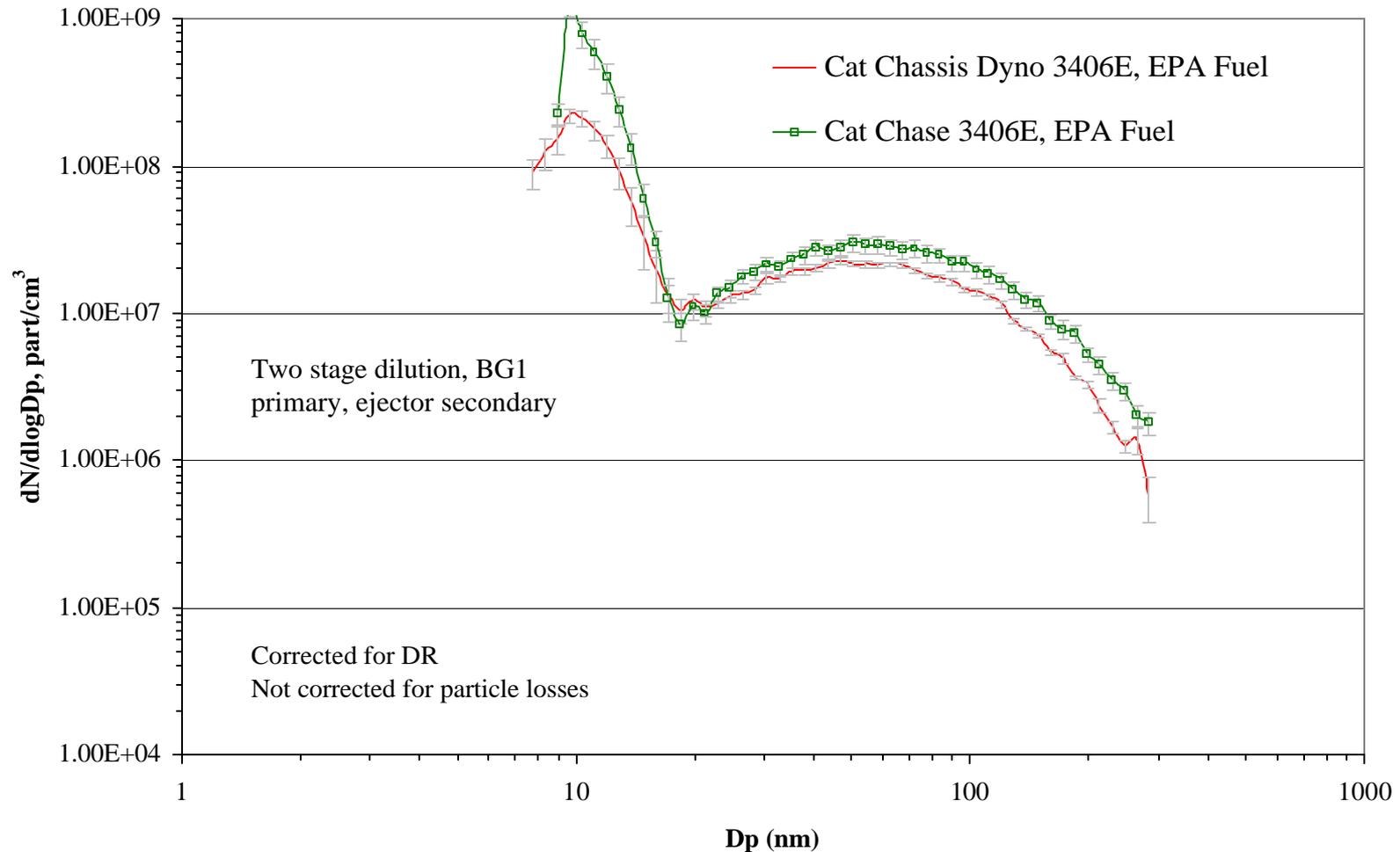
# Composite on-road chase results show much less scatter. Character of size distribution from current and older technology similar.

Composite Graphs: Cat Chase

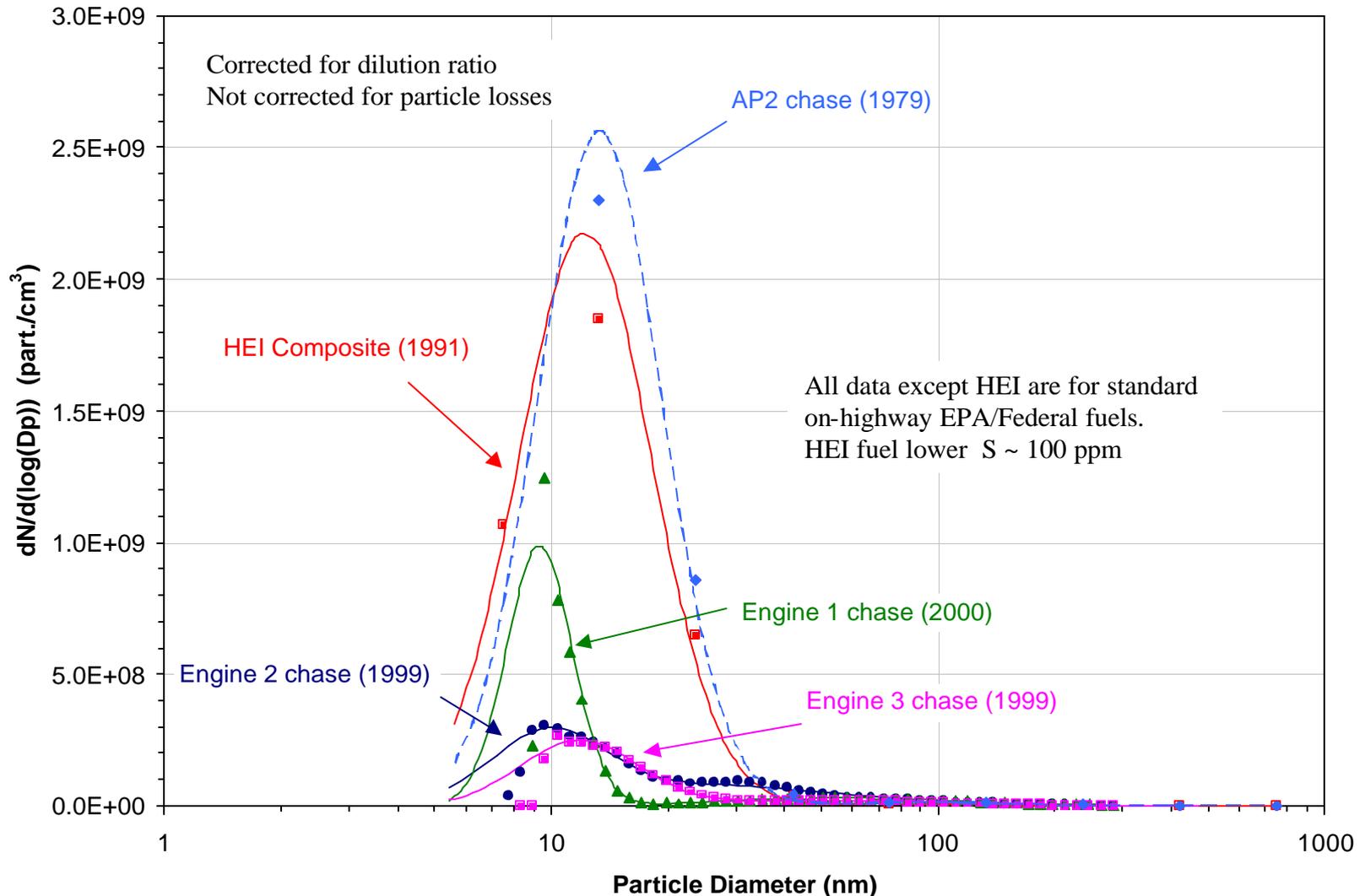


# Comparison of lab and chase measurements – Typical composite results, EPA fuel

Composite Graphs: Cat CD, 3406E, EPA, BG1 Vs. Chase



# Comparison with previous studies: Nanoparticles from newer engines are at lower concentrations and somewhat smaller



# E-43 Questions and answers

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- Can we make laboratory size distribution measurements that mimic real world measurements?
  - *On-road results are very dependent upon dilution conditions like ambient temperature and previous operating history – what condition are we trying to mimic?*
  - *However, we found that although laboratory results are also extremely sensitive to sampling and dilution conditions, we could design systems that give results similar to on-road composite highway cruise and acceleration conditions measured under moderate summer conditions (20-30 C).*
- Do modern Diesel engines produce nanoparticles under real world dilution conditions?
  - *Yes and so do mixed on-road fleets, even in the absence of significant Diesel traffic.*
  - *Nuclei mode formation strongly dependent on ambient temperature and traffic conditions.*
- Do new low carbon emitters produce more nanoparticles than older designs?
  - *No substantial difference has been observed for engines tested in E-43.*
  - *Nuclei mode formation linked to volatile precursor (hydrocarbon and sulfuric acid) concentrations, especially under on-road conditions*

# E-43 Questions and answers

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- What the chemical and physical characteristics of the nanoparticles?
  - *Previous talk showed that they consist mainly volatile materials like heavy hydrocarbons, sulfuric acid, and ...*
  - *No evidence of **significant** solid fraction.*
- How long do they persist in the atmosphere?
  - *Modeling (Capaldo and Pandis, 2002) indicates that for typical urban conditions characteristic times and transit distances for 90% reduction of total number (mainly ultrafine) concentrations are on the order of a few minutes and 100-1000 m, respectively.*
  - *Thus high ultrafine and nanoparticle concentrations from engines are expected to be found mainly on and near roadways – a hotspot problem.*

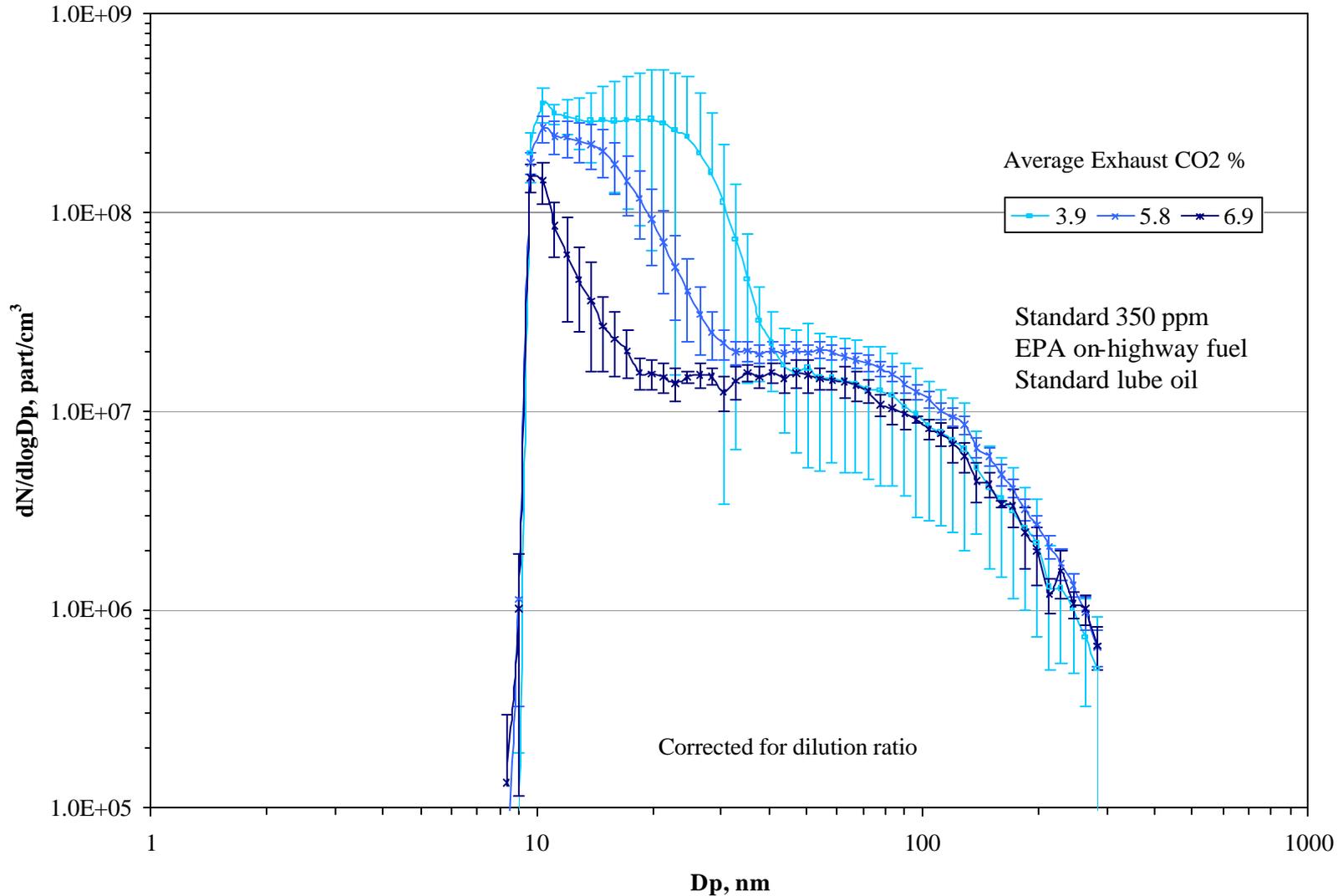
# Sniffing our own exhaust – some new results

Sponsored by Johnson-Matthey, BP/Amoco, Castrol, Corning, Volvo

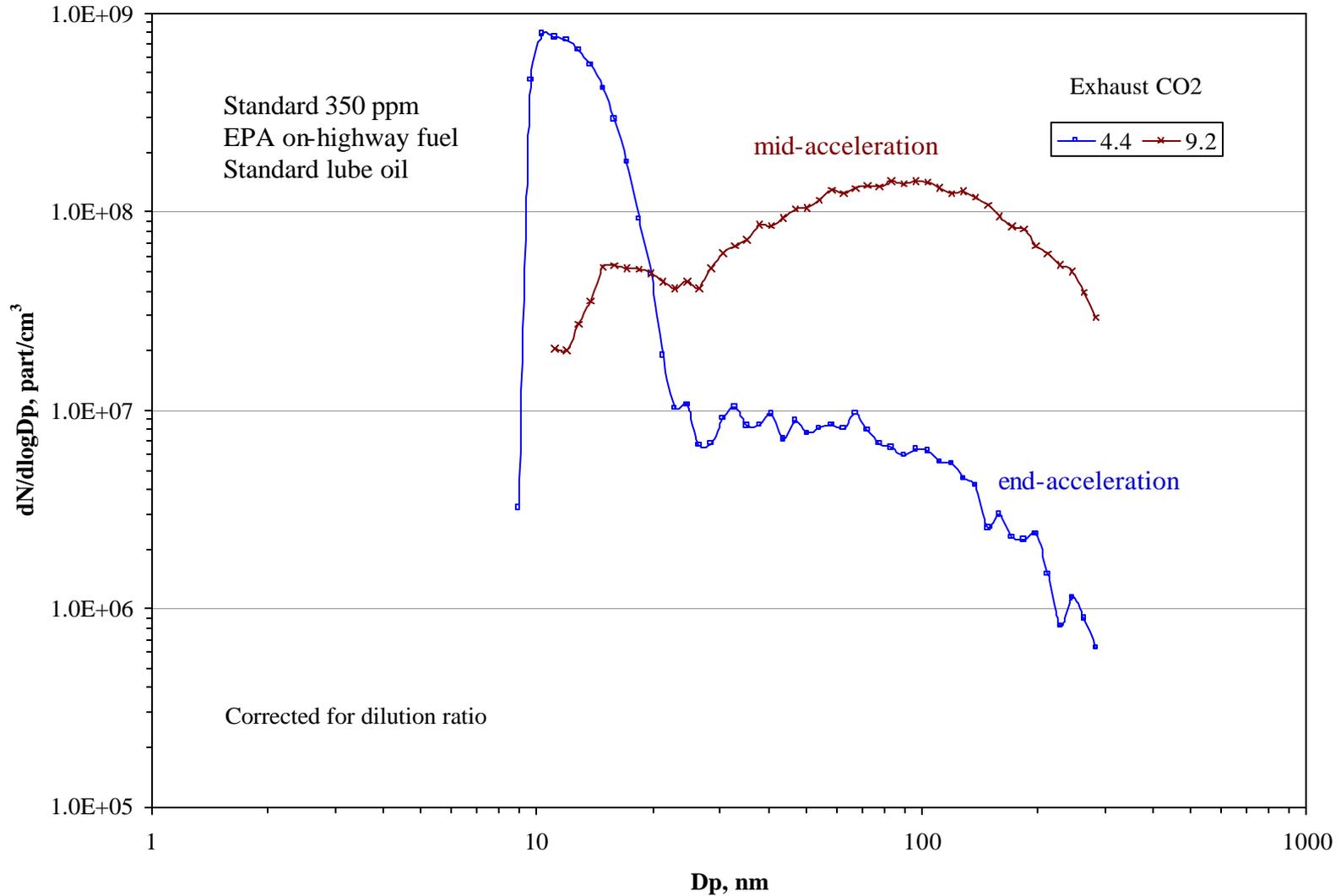
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# Volvo on-road cruise, variable load as indicated by CO2



# Hard acceleration produces large accumulation mode, hot overrun a large nuclei mode



# Ongoing plume sniffing work with mobile lab

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- 15 ppm S fuel, reduced S lube oil, no aftertreatment
- 15 ppm S fuel, reduced S lube oil, CRT (CSF)
- 50 ppm S fuel, reduced S lube oil, CRT (CSF)
- 50 ppm S fuel. Standard S lube oil, CRT (CSF)
- 50 ppm S fuel. Standard S lube oil, no aftertreatment
- NO<sub>x</sub> aftertreatment??

# Continuous instruments show additional structure

